



Bioorganic & Medicinal Chemistry Letters

Bioorganic & Medicinal Chemistry Letters 14 (2004) 4705-4710

## Long-chain formoterol analogues: an investigation into the effect of increasing amino-substituent chain length on the $\beta_2$ -adrenoceptor activity

Vahid Alikhani, David Beer, David Bentley, Ian Bruce, Bernard M. Cuenoud, Robin A. Fairhurst,\* Peter Gedeck, Sandra Haberthuer, Claire Hayden, Diana Janus, Lynne Jordan, Christine Lewis, Kirsty Smithies and Elke Wissler

Novartis Horsham Research Centre, Wimblehurst Road, Horsham, West Sussex RH12 5AB, United Kingdom

Received 24 May 2004; revised 21 June 2004; accepted 25 June 2004 Available online 23 July 2004

Abstract—The synthesis of a series of long-chain formoterol analogues in which the terminal ether residue of the  $\beta$ -phenethylamino-substituent has been extended beyond the methyl ether residue present in the parent compound are described. Evaluation of these analogues as  $\beta_2$ -adrenoceptor agonists was used to provide an insight into the factors controlling the magnitude and duration of receptor activation.

© 2004 Elsevier Ltd. All rights reserved.

The two currently prescribed inhaled long-acting  $\beta_2$ -adrenoceptor agonists formoterol 1 and salmeterol 2 play a central role as bronchodilators in the treatment of the respiratory diseases asthma,  $^1$  and COPD. $^2$  Formoterol is characterised as a high-efficacy agonist with a rapid onset of action. $^3$  In contrast, salmeterol is a low-efficacy agonist with a relatively slow onset of action. $^4$  Despite these differences both compounds provide sustained relief of bronchoconstriction following a twice-daily dosing regimen. In common with others, we are interested in identifying inhaled  $\beta_2$ -agonists with extended durations of action, which are suitable for once-daily dosing as the next generation of compounds from this therapeutic class. $^5$ 

To rationalise the duration of action of formoterol and salmeterol two hypothese have been put forward involving a key interaction with either the receptor, termed the exosite theory,  $^6$  or the surrounding membrane, termed the diffusion microkinetic theory. In the former, a secondary interaction distant from the primary agonist binding region is implicated in which the long-chain of salmeterol anchors to an additional portion of the  $\beta_2$ -adrenoceptor, termed the exosite. This interaction is be-

lieved to enable repeated stimulation to occur without complete dissociation of the ligand, leading to the sustained effect. Key to this interaction is the location of the ether residue within the 6-(4-phenylbutoxy)hexyl amino chain of salmeterol, which is thought to provide the conformational freedom for the dual binding modes to occur.<sup>4,8</sup> In the diffusion microkinetic theory, the basic centre and lipophilic nature of both formoterol and salmeterol allows them to partition effectively into the lipid bilayers of smooth muscle following local delivery to the lung. This accumulation of material within the lipid bilayer functions as a depot to maintain an effective concentration of agonist over time, resulting in the observed duration of effect. Additionally for this theory, the greater lipophilicity and nature of the membrane interaction for salmeterol have been proposed to account for the slow onset of action due to entry into the  $\beta_2$ -adrenoceptor directly from the lipid bilayer as the primary route. Some controversy surrounds these two hypothese centering around the role and extent to which the exosite interaction with salmeterol defines the observed pharmacology. 10,11 However, an interaction with this region of the receptor for salmeterol appears to be well established, through natural, 12 and engineered mutations, 13 even if the nature and degree to which the exosite interaction contributes to the molecules biological activity remains to be fully defined.

<sup>\*</sup> Corresponding author. Tel.: +44-(0)-1403-323508; fax: +44-(0)-1403-323307; e-mail: robin.fairhurst@pharma.novartis.com

Although the six-atom nitrogen-to-oxygen separation is believed to be an essential structural element for the activity observed with salmeterol, 4 the hexyl spacer does allow for significant conformational freedom. A consequence of this is to make any accurate prediction of the active spatial orientation in this region of the molecule difficult. Our interest centred on the observation that this proposed key six-atom separation between the basic nitrogen and ether oxygen is also present in a number of other well described β<sub>2</sub>-adrenoceptor agonists, which include picumeterol 3, sibenadet 4, formoterol 1, fenoterol 5 and TA-2005 6.<sup>11,14</sup> The last three of which all incorporate a para-oxygen substituted β-phenethyl-amino-substituent as the common six-atom spacer. Therefore, potentially enabling an equivalent interaction for the common oxygen atom of the ether side chains, as highlighted in Figure 1. The reduced conformational mobility of the β-phenethyl moiety possibility offering the opportunity of a more energetically favourable interaction.

Based upon the above correlation, we hypothesised that the potential exists with formoterol analogues for enabling the same secondary receptor binding, put forward for salmeterol. This we anticipated could be achieved by extension of the para- $\beta$ -phenethyl terminal ether residue, to facilitate an interaction with the exosite region of the  $\beta_2$ -adrenoceptor. Further support for such a commonality of binding modes with the above catecholamine-based agonists can be derived from receptor mutagenesis

in combination with molecular modelling studies. The key interactions are highlighted for salmeterol and formoterol in Figure 2. In summary, the important interactions between agonists and the  $\beta_2$ -adrenoceptor have been established as involving; the catechol mimic and serine residues 204 and 207 on the fifth transmembrane spanning domain (TM5),15 the protonated amine and Asp-113 on TM3, 16 and the benzylic alcohol with the chirally discriminating Asn-293 on TM6.<sup>17</sup> In addition to these primary interactions, Tyr-308 on TM7 has been identified from molecular modelling studies as defining a limiting-surface of the binding pocket for the aminosubstituents of formoterol, salmeterol and TA-2005.<sup>18</sup> This interaction in the case of salmeterol enables the much larger 4-phenylbutyl ether substituent to orientate itself, facilitated by the presence of the ether linkage, back into the  $\beta_2$ -adrenoceptor.<sup>8</sup> A consequence of this alignment is to bring the 4-phenylbutyl ether moiety into an interaction with the amino acids localised around positions 149-158 on TM4, which define the key residues of the proposed exosite.<sup>13</sup>

In this letter we describe our initial results from sequentially increasing the size of the  $\beta$ -phenethyl oxygen substituent for a series of formoterol analogues to match the equivalent terminal ether residue of salmeterol. We anticipated that determination of the  $\beta_2$ -adrenoceptor activity for this series of homologues would provide an insight into the possibility of an analogous interaction to that proposed for salmeterol with the exosite region.

Figure 1. Comparison of  $\beta_2$ -adrenoceptor agonist structures of formoterol, salmeterol, picumeterol, sibenadet, fenoterol and TA-2005.

Figure 2. Key  $\beta_2$ -adrenoceptor/agonist interactions for salmeterol and formoterol.

Table 1. Structures and position of the  $\beta$ -phenethyl ether substituents R

Compound	Substituent: R Regiochemistr	
a	–Ph	para
b	-CH <sub>2</sub> CH <sub>2</sub> Ph	para
c	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> Ph	para
d	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> Ph	para
e		para
f	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> Ph	meta
g	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OPh	meta

In addition a selection of closely related derivatives were prepared and studied to further investigate the factors controlling the magnitude and duration of the effect at the  $\beta_2$ -adrenoceptor.

To explore the above possibility a series of N-benzylated  $\beta$ -phenylethylamines bearing para-substituted ether residues of increasing chain length and terminated with a phenyl moiety were prepared as key intermediates, entries  $\mathbf{9a}$ - $\mathbf{d}$  in Table 1. Additionally, as a way to further explore the structure–activity relationship, related examples were prepared by varying the nature and position of the ether residue, entries  $\mathbf{9e}$ - $\mathbf{g}$  in Table 1. The synthetic sequence used to prepare the intermediate amines  $\mathbf{9b}$ - $\mathbf{g}$  is shown in Scheme 1. Starting from the appropriately

substituted hydroxyphenylacetic acids 7, carbodiimide couplings yielded the corresponding *N*-benzylated carboxamides 8. The phenolic residues were then alkylated with in situ generated alkyl iodides or, in the case of the 2-indanyl example, via a Mitsunobu reaction. Subsequent reduction of the amide functionality, with either diborane or lithium aluminium hydride, gave the desired secondary amines 9b–g. To prepare the bis-arylether intermediate 9a an alternative sequence was employed. Starting from commercial 4-phenoxy-β-phenylethylamine, *N*-benzoylation followed by amide reduction with lithium aluminium hydride produced the desired benzylated amine 9a in 48% overall yield.

The above secondary benzylated amines **9a**–**g** were then reacted under conditions of either conventional or microwave heating to open the chiral epoxide **10**. Reaction occurring selectively at the least hindered position, yielding the ethanolamine derivatives **11a**–**g**. <sup>19</sup> Subsequent functional group manipulation, via hydrogenation of the nitro residues over platinum oxide, followed by formylation of the liberated aniline functionality gave the corresponding formamides. Finally, hydrogenolysis of the benzyl protecting groups yielded the target molecules **12a**–**g**, as single enantiomers of the more active (*R*)-configuration (Scheme 2). <sup>20</sup>

Compounds 12a-g were screened in a filtration binding assay to determine their affinities for the human

**Scheme 1.** Synthesis of the amines **9b–g**: (i) BnNH<sub>2</sub>, EDCI, HOAt, CH<sub>2</sub>Cl<sub>2</sub>, rt (88–93%); (ii) R–Cl, CsCO<sub>3</sub>, KI, CH<sub>3</sub>CN, 90°C (86–90%); or, R–OH, PPh<sub>3</sub>, DEAD, THF, rt (26%); (iii) B<sub>2</sub>H<sub>6</sub>, THF, reflux, then MeOH, reflux (44–95%); or, LiAlH<sub>4</sub>, THF, reflux (29–33%).

Scheme 2. Synthesis of the  $\beta_2$ -adrenoceptor agonists 12a–g: (i) 9a–g toluene, reflux, 18h (53–82%); or, neat, Prolabo Synthewave <sup>TM</sup> 402 microwave, 25 min at 110 °C (63–92%); (ii) 1 atm H<sub>2</sub>, PtO<sub>2</sub>, 1:1 toluene: THF, rt (88–98%); (iii) aged mixture HCO<sub>2</sub>H/Ac<sub>2</sub>O, 1:1 toluene-THF, rt (62–86%); or, HCO<sub>2</sub>H, oxalyl chloride, imidazole, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, rt (25%); (iv) 1 atm H<sub>2</sub>, Pd/C, MeOH, rt (58–89%).

 $\beta_2$ -adrenoceptor.<sup>21</sup> Additionally, IC<sub>50</sub> values for the compounds ability to inhibit the electrically induced contraction of superfused guinea-pig trachea were determined as a measure of the functional response at the  $\beta_2$ -adrenoceptor.<sup>22</sup> Biological data and calculated  $\log D_{7.4}$  values,<sup>23</sup> as an assessment of the membrane partitioning capability, are shown in Table 2. Membrane partitioning being a key determinant of the compounds duration of action potential based upon the diffusion microkinetic theory.<sup>7</sup>

Comparison of the affinities of the compounds 12a-g for the  $\beta_2$ -adrenoceptor showed little variability in the  $K_i$  values with the exception of the biphenyl ether 12a, which was slightly reduced. Interestingly, all of the compounds as the more active (R)-enantiomer exhibited decreased affinities for the  $\beta_2$ -adrenoceptor relative to racemic formoterol. A major proportion of this small drop in affinity could potentially be accounted for by the absence of the (R)- $\alpha$ -methyl substituent of the  $\beta$ -

**Table 2.** Biological data and calculated  $\log D_{7.4}$  values for formoterol, salmeterol and compounds 12a–g

Compound	$\beta_2$ -Adrenoceptor $K_i$ (nM)	$\operatorname{clog} D_{7.4}$	Guinea-pig trachea IC <sub>50</sub> (nM) <sup>a</sup>
Formoterol 1	2.6	0.06	1.4
Salmeterol 2	1.0	1.24	15
12a	35.4	1.95	120
12b	7.8	1.67	1.1
12c	5.9	2.35	38
12d	9.2	2.83	71
12e	11.6	2.95	251
12f	8.5	2.83	114
12g	3.3	2.00	164

<sup>&</sup>lt;sup>a</sup> Determined from maximal response after compound present in superfusate for initial  $30 \min (n \ge 3)$ .

phenethyl moiety in compounds 12a–g, when compared with formoterol.<sup>20</sup> These binding data indicate that no marked change in affinity for the  $\beta_2$ -adrenoceptor is apparent for 12d where, based upon the proposed hypothesis the terminal ether residue is in an equivalent position to that present in salmeterol.

The functional consequences of the above interaction with the  $\beta_2$ -adrenoceptor were measured in a guineapig tracheal strip assay.<sup>22</sup> We noted a greater variability in the ability of the compounds 12a-g to agonise the receptor, measured by inhibition of contraction, when compared with their binding affinities. IC<sub>50</sub> values ranged from being comparable with formoterol, to greater than 200-fold less potent. In all instances greater than an 80% inhibition of the electrically-stimulated contraction was achieved at the higher concentrations investigated in the dose response studies (1 µM). Thus, indicating that all the compounds retain a significant degree of the high-efficacy agonist properties associated with formoterol. Comparing the para-substituted isomers: the biphenyl ether 12a in accord with the above binding data resulted in a marked 86-fold fall in functional potency relative to formoterol. Increasing the separation between the phenyl residues by incorporation of an alkyl chain resulted in an improvement in functional activity to the level of formoterol for the para-substituted 2-phenethyl ether derivative 12b, with the functional potency decreasing upon increasing alkyl chain length in going from two to four methylene groups, entries 12b-d. However, a satisfactory level was still retained in the para-4-phenylbutyl ether derivative 12d, with the terminal ether residue equivalent to salmeterol. This decrease in potency of up to 51-fold relative to formoterol may provide a better opportunity for formulation as an inhalation product with activity in this range.<sup>24</sup> The 2-indanyl derivative **12e**, a cyclic analogue of the potent 12b, resulted in a marked 228-fold fall in functional activity although this was again not apparent from the binding data. Therefore, this result potentially indicates that steric bulk and conformational restriction about the aryl ether linkage are relatively poorly tolerated for receptor activation in this series of compounds, a rationale, which is further supported by the reduced activity seen for the biphenyl ether 12a. Switching the point of attachment of the ether linkage in the β-phenethyl moiety to a meta-relationship and hence reducing the six-atom separation between basic nitrogen and ether oxygen by one atom resulted in essentially no change in functional activity, entry 12f, compared with the para-isomer 12d. Thus, these data suggest that the role of the side-chain ether oxygen in this  $\beta$ -phenethyl series is less critical for receptor activation compared with salmeterol. Incorporation of a second ether residue into the four-atom *meta*-linker, entry 12g, was tolerated with little change in activity compared to the carbon analogue 12f.

Comparison of the above functional data with the binding affinities highlights the need for a functional assessment to understand the impact of the above structural changes on receptor activation. The nature and position of the ether residue in the series of compounds 12 having

been shown to have a significant effect on functional  $\beta_2$ adrenoceptor activation. Analysis of the above structure–activity relationship does not highlight a trend for an optimal interaction to indicate that the most favourable profile is attained with 12d, which based upon the hypothesised structural equivalence is proposed as the closest analogue in this series to salmeterol. However, this approach has enabled functionally-active and high-efficacy  $\beta_2$ -adrenoceptor agonists to be identified of increased lipophilicity of up to greater than 2.8 calculated  $\log D_{7.4}$  units higher than formoterol. Based upon the diffusion microkinetic theory, these compounds have the potential to produce the desired increase in duration of action. Representative duration of action data from a superfused guinea-pig tracheal strip assay, in which a drug-free washout phase of up to 9.5h was included, for formoterol, 12d, and 12g, are shown in Figure 3.<sup>22</sup> The concentrations shown for each compound are selected to most closely approximate to an IC<sub>80</sub> at the end of the 30min compound administration phase. Durations of action were taken as the time to return to 50% of the maximal effect for each compound from the end of the 30min compound administration phase and are summarised in Table 3 with measured  $\log D_{7.4}$ , <sup>25</sup> and  $K_{\rm IAM}$  values, <sup>26</sup> as an assessment of the membrane partitioning capability. <sup>27</sup>

The measured data to assess the membrane partitioning capabilities for formoterol, **12d**, and **12g** were found to correlate with the calculated  $\log D_{7.4}$  values presented in Table 2. Taking the differences in the  $K_{\rm IAM}$  values

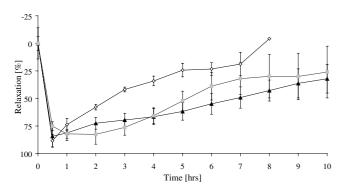


Figure 3. Electrically-stimulated superfused guinea-pig tracheal strip assay (Electrical stimulation was delivered every  $2 \,\mathrm{min}$ , data shown are for each hour following starting the superfusion of the test compound, and at the end of the  $30 \,\mathrm{min}$  drug administration phase. Data are expressed as mean  $\pm$  standard error of the mean) data for formoterol, 12d and  $12g \diamondsuit$  formoterol  $30 \,\mathrm{nM}$  (n=3),  $\blacktriangle$  12d  $100 \,\mathrm{nM}$  (n=5),  $\blacksquare$  12g  $1000 \,\mathrm{nM}$  (n=4).

**Table 3.** Durations of action and measured  $\log D_{7.4}$  values for formoterol, **12d** and **12g** 

Compound	Duration-of-action (h)	$\log D_{7.4}$	$K_{\text{IAM}}$
Formoterol 1	1.92	0.79	0.70
12d	6.62	2.76	>2.73 <sup>a</sup>
12g	5.43	2.49	2.18

 $<sup>^{</sup>a}\log K_{\mathrm{IAM}}$  value above upper limit of measurable range for assay system.

for the long-chain analogues 12d and 12g: a greater than 107-fold and 30-fold increased preference for a phospholipids, over an aqueous phase, respectively, was determined when compared with formoterol. This increased capacity for retention in lipid bilayers was reflected in an increase in the duration of the relaxant effect over formoterol of, 3.4-fold and 2.8-fold, respectively, for 12d and 12g when compared at equi-effective concentrations. More than a doubling of the duration of effect over formoterol being obtained not only for 12d where the hypothesised key six-atom nitrogen-to-oxygen separation is present, but also 12g where it has been reduced by one atom. A one-atom shift in the position of the ether oxygen having previously been shown to have a marked impact on the duration of action for a series of salmeterol analogues.4 Thus, when we assessed the differences seen between 12d and 12g the extent of the increase in lipophilicity appeared to best explain the degree of enhancement in duration of action observed over formoterol rather than as a consequence of a specific interaction with the exosite region of the  $\beta_2$ -adrenoceptor. Additionally, from the data shown in Figure 3, the maximal effect for 12d, like formoterol, is achieved within 5 min of the end of the initial 30 min drug administration phase, which is consistent with the potential for a rapid onset of action to be conserved.<sup>22</sup>

In summary, a series of long-chain formoterol analogues in which the ether residue of the β-phenethyl-amino-substituent was varied have been evaluated as  $\beta_2$ -adrenoceptor agonists to explore if an analogous exosite interaction to that proposed for salmeterol was evident. In vitro studies for this series of compounds highlighted the need for a functional assessment to fully understand the structure-activity relationship for receptor activation. Although evaluation of the data does not rule out an interaction with the exosite region of the receptor for the compounds 12, no variation in the degree and duration of agonist activity to support a dual binding mode was apparent. However, this approach has enabled high-efficacy  $\beta_2$ -adrenoceptor agonists of increased lipophilicity to be identified, which from the diffusion microkinetic theory would be anticipated to have the potential for an extended duration of action as a result of greater membrane partitioning. This has been demonstrated for the compounds 12d and 12g, which produced a greater than 2.5-fold longer duration of action compared to the currently prescribed twice-daily  $\beta_2$ -adrenoceptor agonist formoterol. Additionally in the case of 12d, this increased duration of action was also associated with the desired rapid onset of action profile. Further studies to explore the potential of the long-chain analogue 12d (NVP-QAC455) as a once-daily long-acting  $\beta_2$ -adrenoceptor agonist will be described in due course.

## References and notes

- Global Initiative for Asthma: Global strategy for asthma management and prevention. NHLBI/WHO Workshop Report, Publ 02-3659, 2002.
- 2. Global Initiative for Obstructive Lung Disease: Global strategy for the diagnosis, management and prevention of

- chronic obstructive pulmonary disease. National Heart lung and Blood Institute, Bethesda, 2000.
- 3. Matthys, H. Respiration 2001, 68, 432.
- 4. Johnson, M. Med. Res. Rev. 1995, 15, 225.
- 5. Anon Expert Opin. Ther. Pat. 2003, 13, 273.
- Coleman, R. A.; Johnson, M.; Nials, A. T.; Vardey, C. J. Trends Pharmacol. Sci. 1996, 17, 234.
- Anderson, G. P.; Lindén, A.; Rabe, K. F. Eur. Respir. J. 1994, 7, 569.
- 8. Lewell, X. Q. Drug Des. Discovery 1992, 9, 29.
- Ochsner, M.; Jaekel, K.; Mutz, M.; Anderson, G. P.; John, E. Eur. J. Med. Chem. 1994, 34, 451.
- (a) Teschemacher, A.; Lemoine, H. *J. Pharmacol. Exp. Ther.* **1999**, *288*, 1084; (b) Bergendal, A.; Lindén, A.; Skoogh, B. E.; Gerspacher, M.; Anderson, G. P.; Löfdahl, C. G. *Br. J. Pharmacol.* **1996**, *117*, 1009.
- 11. Austin, R. P.; Barton, P.; Bonnert, R. V.; Brown, R. C.; Cage, P. A.; Cheshire, D. R.; Davis, A. M.; Dougall, I. G.; Ince, F.; Pairaudeau, G.; Young, A. *J. Med. Chem.* **2003**, *46*, 3210.
- Green, S. A.; Rathz, D. A.; Schuster, A. J.; Liggett, S. B. Eur. J. Pharmacol. 2001, 421, 141.
- Green, S. A.; Spasoff, A. P.; Coleman, R. A.; Johnson, M.; Liggett, S. B. J. Biol. Chem. 1996, 271, 24029.
- 14. Waldeck, B. Eur. J. Pharmacol. 2002, 445, 1.
- Strader, C. D.; Candelore, M. R.; Rands, E.; Hill, W. S.;
  Dixon, R. A. F. J. Biol. Chem. 1988, 263, 10267.
- Strader, C. D.; Candelore, M. R.; Hill, W. S.; Sigal, I. S.; Dixon, R. A. F. J. Biol. Chem. 1989, 264, 13572.
- Wieland, K.; Zuurmond, H. M.; Krasel, C.; IJzerman, A. P.; Lohse, M. J. *Proc. Natl. Acad. Sci. U.S.A.* 1996, 93, 9726.

- (a) Furse, K. E.; Lybrand, T. P. J. Med. Chem. 2003, 46, 4450;
  (b) Isogaya, M.; Sugimoto, Y.; Tanimura, R.; Tanaka, R.; Kikkawa, H.; Nagao, T.; Kurose, H. Mol. Pharmacol. 1999, 56, 875;
  (c) Isogaya, M.; Yamagiwa, Y.; Fujita, S.; Sugimoto, Y.; Nagao, T.; Kurose, H. Mol. Pharmacol. 1998, 54, 616;
  (d) Kikkawa, H.; Isogaya, M.; Nagao, T.; Kurose, H. Mol. Pharmacol. 1998, 53, 128.
- Hett, R.; Fang, Q. K.; Gao, Y.; Hong, Y.; Butler, H. T.;
  Nie, X.; Wald, S. A. Tetrahedron Lett. 1997, 38, 1125.
- (a) Waldeck, B. Chirality 1993, 5, 350; (b) Trofast, J.;
  Österberg, K.; Källström, B.-L.; Waldeck, B. Chirality 1991, 3, 443; (c) Murase, K.; Mase, T.; Ida, H.; Takahashi, K.; Murakami, M. Chem. Pharm. Bull. 1978, 26, 1123.
- 21.  $K_i$  values were determined by the displacement of  ${}^3H$  CGP12177 from the human  $\beta_2$ -adrenoceptor expressed in insect Sf21 membranes.
- Nials, A. T.; Sumner, M. J.; Johnson, M.; Coleman, R. A. J. Pharmacol. 1993, 108, 507.
- 23. Calculated log *D*<sub>7,4</sub> values were generated using ACD cLogD version 7.06 software.
- Malcolmson, R. J.; Embleton, J. K. *Pharm. Sci. Tech. Today* 1998, 1, 394.
- 25. log *D*<sub>7,4</sub> values were measured by potentiometric titration, using a GlpKa instrument from Sirius Analytical Instruments (Forest Row, UK).
- 26.  $\log K_{\rm IAM}$  values were determined from HPLC retention times, using a  $10\times3\,\mathrm{mm}$  IAM fast screen mini column (Regis speciality columns) with isocratic eluent;  $1\,\mathrm{mL\,min}^{-1}$ , 85% mixed phosphate pH7.4 buffer, 15% acetonitrile and UV detection at 230 nm.
- 27. Taillardat-Bertschinger, A.; Carrupt, P. A.; Barbato, F.; Testa, B. J. Org. Chem. 2003, 46, 655.